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# Full Papers

## New Lanostanoid Glycosides from the Fruit Body of Laetiporus versisporus

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Four new lanostanoid glycosides,  $3\beta$ ,  $7\alpha$ -dihydroxy-24-methylene-lanost-8-en-21-oic acid 7-O- $\beta$ -D-glucopyranoside (**1**),  $3\beta$ ,  $7\alpha$ -dihydroxy-lanost-8,24-dien-21-oic acid 7-O- $\beta$ -D-glucopyranoside (**2**),  $7\alpha$ -hydroxy-3-ketolanost-8,24-dien-21-oic acid 7-O- $\beta$ -D-glucopyranoside (**3**), and  $7\alpha$ -hydroxy-3-keto-24-methylene-lanost-8en-21-oic acid 7-O- $\beta$ -D-glucopyranoside (**4**) were isolated from the fruit bodies of *Laetiporus versisporus*. Compounds **1**–**4** were named laetiposides A–D, respectively. Their structures were established by extensive NMR experiments and chemical methods.

Laetiporus versisporus (Polyporus calvatioides) (Polyporaceae) grows on the dead trees in broad-leaf forests and is distributed throughout Japan.<sup>1,2</sup> A phytochemical study of the genus of *Laetiporus* (*Polyporus*) reported the isolation of polyporusterones  $A-G^3$  and  $9\alpha$ -hydroxy-1,2,3,4,5,10,19-heptanorergosta-7,22-diene-6,9-lactone.<sup>4</sup> However, a systematic investigation of *L. versisporus* has not been reported. Therefore, it appeared of interest to investigate the high polar components of this fungus.

Four new lanostanol glycosides, laetiposides A (1), B (2), C (3), and D (4), along with eburicoic acid (5),<sup>5</sup> trametenolic acid (6),<sup>5</sup> sulfurenic acid,<sup>6</sup> and  $15\alpha$ -dihydroxytrametenolic acid<sup>6</sup> were isolated from the EtOH extract of the fruit bodies of *L. versisporus* by column chromatography and reversed-phase HPLC. We describe here the isolation and structure elucidation of laetiposides A–D (1–4) by various NMR techniques, including COSY, HMQC, HMBC, and ROESY experiments, and chemical degradation.

## **Results and Discussion**

The main compound, laetiposide A (1), was obtained as a colorless needle, which gave a molecular ion peak at m/z 647 in its FABMS. This corresponds to a molecular formula  $C_{37}H_{60}O_{9}$ , requiring eight unsaturation equivalents. The IR spectrum of 1 showed broad absorption at 1700 cm<sup>-1</sup>

(carboxyl). The 37 carbon signals were observed in the <sup>13</sup>C NMR spectrum; these were sorted, by DEPT experiment, into seven methyl carbons; nine methylene carbons; six methine carbons, two of which had oxygen substituents; four sp<sup>2</sup> carbons, one of which was protonated; four sp<sup>3</sup> quaternary carbons; and a carboxyl carbon, in addition to the six signals of a hexose (Table 1). These data suggested 1 to be a tetracyclic triterpene containing two double-bonds  $(\delta 156.0, 142.1, 134.5, and 107.0)$ , one carboxyl ( $\delta 178.8$ ), two secondary hydroxy moieties ( $\delta$  77.8 and 72.2), and one glucopyranosyl moiety ( $\delta$  101.2, 78.5, 78.3, 74.6, 72.4, and 63.6). A combination of COSY and HMBC experiments enabled us to construct a 24-methylenelastanol skeleton as the aglycon of 1. A carboxyl group was located at the C-21 position from HMBC correlation of H-20 at  $\delta$  2.64 to C-21 (Figure 1). A <sup>13</sup>C NMR spectral comparison of 1 with eburicoic acid (5)<sup>5</sup> showed that 1 differs structurally from 5 only in the B ring. One of two secondary hydroxy carbons ( $\delta$  72.2) was defined at C-7 from HMBC correlations of H-7 ( $\delta$  4.66, br s) to C-5 ( $\delta$  44.3), C-8 ( $\delta$  134.5), and C-9 ( $\delta$  142.1) (Figure 1). Further, the long-range coupling between the H-1 ( $\delta$  4.97) of the glucosyl unit and C-7 ( $\delta$  72.2) established that the glucosyl group is attached to C-7. The  $\alpha$ -O function at C-7 position could be assigned from the broad singlet signal observed for H-7 and the NOE experiment (Figure 2). The acid hydrolysis of 1 with 5%  $H_2SO_4$  afforded dehydroeburicoic acid  $(7)^5$  and D-glucose, which was confirmed by specific rotation using chiral detection by HPLC

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analysis.<sup>7,8</sup> The coupling constant for the anomeric protons observed at  $\delta$  4.97 (J = 7.0 Hz) in the <sup>1</sup>H NMR spectrum of **1** indicated the glucose had a  $\beta$  configuration. Thus, the structure of **1** was formulated as  $3\beta$ , $7\alpha$ -dihydroxy-24-methylene-lanost-8-en-21-oic acid 7-O- $\beta$ -D-glucopyranoside.

Laetiposide B (2) gave a quasi molecular ion at m/z 633  $[M - H]^-$ , 14 mass units (CH<sub>2</sub>) less than that of **1**. Acid hydrolysis of **2** also yielded D-glucose, which has the  $\beta$  configuration by its coupling constant (J = 7.8 Hz) of the anomeric proton at  $\delta$  4.98. Comparison of physicochemical data of **2** with those of **1** revealed that the only difference was that **2** had a hydrogen at C-24 instead of a methylene. The carbon signals due to the side-chain part of **2** were in good agreement with those of trametenolic acid (**6**).<sup>5</sup> The connectivities of the COSY and HMBC experiments supported the assumed structure of **2**. Accordingly, **2** was formulated as  $3\beta$ ,  $7\alpha$ -dihydroxy-lanost-8,24-dien-21-oic acid 7-O- $\beta$ -D-glucopyranoside.

The FABMS of laetiposide C (**3**) gave a quasi molecular ion at m/z 631 [M – H]<sup>-</sup>, 2 mass units less than that of **2**. A <sup>13</sup>C NMR spectral comparison of **3** with **2** showed that **3** differs structurally from **2** only in the A ring, namely, the presence of a carbonyl function at C-3 in **3** instead of a hydroxyl group in **2**. Indeed, in the <sup>13</sup>C NMR spectrum of **3**, a signal at  $\delta$  216.7 was observed. Long-range correlations, in the HMBC spectrum of **3**, observed for H<sub>2</sub>-2 ( $\delta$ 2.12 and 2.31), H-28 ( $\delta$  1.36), H-29 ( $\delta$  1.11), and C-3 ( $\delta$ 216.7) indicated a carbonyl function at the C-3 position. Accordingly, **3** was formulated as 7 $\alpha$ -hydroxy-3-keto-lanost-8,24-dien-21-oic acid 7-*O*- $\beta$ -D-glucopyranoside.

The last compound, laetiposide D (4) showed the molecular formula  $C_{37}H_{58}O_9$  from the elemental analysis data and the observation of 37 carbon signals in the <sup>13</sup>C NMR spectrum. Comparison of physicochemical data of 4 with those of 1 revealed that the only difference was that 4 had a carbonyl function at C-3 instead of a hydroxy group. Therefore, 4 was formulated as  $7\alpha$ -hydroxy-3-keto-24-methylene-lanost-8-en-21-oic acid 7-*O*- $\beta$ -D-glucopyranoside.

Table 1.  $^{13}\text{C}$  NMR Data for Compounds 1–4 and 6 (in  $C_5D_5N$  at 100 MHz)

position	1	2	3	4	6
C-1	35.4 t	35.3 t	35.2 t	35.1 t	36.1 t
2	28.5 t	28.5 t	34.2 t	34.2 t	28.7 t
3	77.8 d	77.7 d	216.7 s	216.8 s	78.0 d
4	39.1 s	39.1 s	46.8 s	46.8 s	39.5 s
5	44.3 d	44.2 d	43.9 d	43.8 d	50.9 d
6	23.6 t	23.4 t	24.7 t	24.7 t	18.7 d
7	72.2 d	72.0 d	71.9 d	71.9 d	26.6 t
8	134.5 s	134.5 s	135.1 s	135.1 s	135.2 s
9	142.1 s	142.1 s	140.2 s	140.2 s	134.6 s
10	38.5 s	38.5 s	38.0 s	37.9 s	37.4 s
11	21.6 t	21.4 t	21.3 t	21.2 t	21.3 t
12	29.6 t	29.4 t	29.3 t	29.2 t	29.4 t
13	45.6 s	45.4 s	45.4 s	45.4 s	44.9 s
14	50.4 s	50.2 s	50.4 s	50.3 s	49.9 s
15	29.4 t	29.2 t	29.5 t	29.5 t	30.9 t
16	27.8 t	27.6 t	27.9 t	27.6 t	27.5 t
17	47.8 d	47.7 d	47.8 d	47.8 d	47.7 d
18	16.6 q	16.5 q	16.7 q	16.6 q	16.3 q
19	18.0 q	17.8 q	17.7 q	17.7 q	19.4 q
20	49.5 d	49.3 d	49.3 d	49.4 d	49.0 d
21	178.8 s	178.8 s	178.7 s	178.8 s	178.6 s
22	31.8 t	33.2 t	33.3 t	31.7 t	33.3 t
23	32.7 t	26.6 t	26.7 t	32.7 t	26.7 t
24	156.0 s	124.9 d	124.9 d	155.9 s	124.9 d
25	34.2 d	131.6 s	131.7 s	34.2 d	131.6 s
26	21.9 q	25.6 q	25.8 q	21.9 q	25.8 q
27	22.0 q	17.6 q	17.7 q	22.0 q	17.7 q
28	28.6 q	28.5 q	27.2 q	27.2 q	28.6 q
29	16.9 q	16.7 q	21.3 q	21.2 q	16.4 q
30	25.7 q	25.6 q	25.5 q	25.5 q	24.5 q
31	107.0 t			107.0 t	
glc C-1	101.2 d	101.1 d	101.7 d	101.7 d	
2	74.6 d	75.2 d	75.1 d	75.1 d	
3	78.5 d	78.4 d	78.5 d	78.5 d	
4	72.4 d	72.4 d	72.4 d	72.4 d	
5	78.3 d	78.2 d	78.3 d	78.3 d	
6	63.6 t	63.6 t	63.5 t	63.5 t	



Figure 1. COSY and selected HMBC correlations of laetiposide A (1).



Figure 2. Selected NOE correlations of laetiposide A (1).

### **Experimental Section**

**General Experimental Procedures.** Melting points were measured with a Yanagimoto micromelting point apparatus and were uncorrected. Optical rotations were obtained on a JASCO DIP-360 polarimeter. IR spectra were recorded on a JASCO FT/IR-5300. NMR spectra were recorded on a JEOL GSX-400 spectrometer in pyridine-*d*<sub>5</sub>, using TMS as internal standard. NMR experiments included <sup>1</sup>H-<sup>1</sup>H COSY, HMQC, HMBC, and ROESY. Coupling constants (*J* values) are given in Hz. The FABMS (Xe gun, 10 kV, triethylene glycol as the matrix) was measured on a JEOL JMS-PX303 mass spectrometer. HPLC separations were performed with a Hitachi HPLC system (L-7100 pump, L-4000 UV).

**Plant Material.** The fruit bodies of *Laetiporus versisporus* were collected at Tokushima City, Japan, in autumn 1997. A voucher specimen (TB3001) is deposited in the Herbarium of Faculty of Pharmaceutical Sciences, Tokushima Bunri University, Tokushima, Japan.

**Extraction and Isolation.** The fresh fruit bodies (1.3 kg) of L. versisporus were extracted with 70% EtOH at room temperature for 6 weeks. The EtOH extract (57.5 g) was partitioned between H<sub>2</sub>O and EtOAc. The EtOAc-soluble portion (9.0 g) was repeatedly subjected to Si gel column chromatography with CH<sub>2</sub>Cl<sub>2</sub>-MeOH-H<sub>2</sub>O (25:1:0-25:4:0.1) to afford five fractions (fractions 1-5). Fractions 2 (3.0 g) and 3 (1.0 g) were purified by preparative HPLC (ODS, 70-72%CH<sub>3</sub>OH) to afford eburicoic acid (5, 135 mg), trametenolic acid (6, 65 mg), sulfurenic acid (15 mg), and 15 $\alpha$ -dihydroxytrametenolic acid (30 mg). Fraction 5 (2.0 g) was further subjected to Si gel column chromatography with CH2Cl2-MeOH-H2O (25:4:0.1) to give four fractions fractions 5-1-4). Fractions 5-2(0.2 g) and 5-3 (0.21 g) were purified by preparative HPLC (ODS, 68% CH<sub>3</sub>OH) to afford laetiposides C (3, 20 mg) and D (4, 35 mg) from fraction 5-2, and laetiposides A (1, 62 mg) and B (2, 10 mg) from fraction 5-3.

**Laetiposide A (1):** colorless needles, mp  $276-277 \,^{\circ}$ C;  $[\alpha]^{25}_{D}$  +9.9° (*c* 2.0, MeOH); FT–IR (dry film)  $\nu_{max}$  3400 (OH), 1700 (C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR (C<sub>5</sub>D<sub>5</sub>N, 400 MHz)  $\delta$  0.98 (3H, s, Me-19), 1.01 (3H, s, Me-18), 1.03 (6H, d, J = 6.3 Hz, Me-26 and 27), 1.09 (3H, s, Me-29), 1.40 (3H, s, Me-30), 1.46 (3H, s, Me-28), 2.28 (1H, m, H-25), 2.50 (1H, q, J = 9.9 Hz, H-17), 2.64 (1H, td, J = 11.0, 3.0 Hz, H-20), 3.13 (1H, dd, J = 11.3, 4.0 Hz, H-3), 4.66 (1H, br s, H-7), 4.97 (1H, d, J = 7.0 Hz, H-1 of glc), 4.89 and 4.92 (each 1H, br s, H<sub>2</sub>-31); <sup>13</sup>C NMR (C<sub>5</sub>D<sub>5</sub>N), see Table 1; FABMS m/z [M – H] – 647, [M – H – glc – H<sub>2</sub>O] – 467; *anal.* C 68.77%, H 9.13%, calcd for C<sub>37</sub>H<sub>60</sub>O<sub>9</sub>, C 68.49%, H 9.32%.

**Laetiposide B (2):** colorless needles, mp 232–234 °C;  $[\alpha]^{25}_{\rm D}$ +12.8° (*c* 0.7, MeOH); FT–IR (dry film)  $\nu_{\rm max}$  3335 (OH), 1700 (C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR (C<sub>5</sub>D<sub>5</sub>N, 400 MHz)  $\delta$  0.98 (3H, s, Me-19), 1.03 (3H, s, Me-18), 1.09 (3H, s, Me-29), 1.38 (3H, s, Me-30), 1.46 (3H, s, Me-28), 1.62 (3H, s, Me-27), 1.68 (3H, s, Me-26), 2.48 (1H, q, J = 9.9 Hz, H-17), 2.65 (1H, td, J = 11.0, 3.0 Hz, H-20), 3.14 (1H, dd, J = 11.3, 4.0 Hz, H-3), 4.67 (1H, br s, H-7), 4.98 (1H, d, J = 7.8 Hz, H-1 of glc), 5.33 (1H, m, H-24); <sup>13</sup>C NMR (C<sub>5</sub>D<sub>5</sub>N), see Table 1; FABMS *m*/*z* [M – H] <sup>-</sup> 633, [M – H – glc – H<sub>2</sub>O] <sup>-</sup> 453; *anal.* C 67.97%, H 9.56%, calcd for C<sub>36</sub>H<sub>58</sub>O<sub>9</sub>, C 68.11%, H 9.21%.

**Laetiposide C (3):** colorless needles, mp 179–181 °C;  $[\alpha]^{25}_{\rm D}$  +29.1° (*c* 1.9, MeOH); FT–IR (dry film)  $\nu_{\rm max}$  3380 (OH), 1700 (C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR (C<sub>5</sub>D<sub>5</sub>N, 400 MHz)  $\delta$  0.89 (3H, s, Me-19), 1.02 (3H, s, Me-18), 1.11 (3H, s, Me-29), 1.36 (3H, s, Me-28), 1.37 (3H, s, Me-30), 1.62 (3H, s, Me-27), 1.68 (3H, s, Me-26), 2.46 (1H, q, J = 9.9 Hz, H-17), 2.64 (1H, td, J = 11.0, 3.0 Hz, H-20), 4.58 (1H, br s, H-7), 4.92 (1H, d, J = 7.7 Hz, H-1 of

glc), 5.34 (1H, m, H-24); <sup>13</sup>C NMR ( $C_5D_5N$ ), see Table 1; FABMS m/z [M – H] <sup>-</sup> 631, [M – H – glc – H<sub>2</sub>O] <sup>-</sup> 451; anal. C 68.20%, H 8.78%, calcd for  $C_{36}H_{56}O_9$ , C 68.33%, H 8.92%. **Laetiposide D (4)**: colorless needles, mp 232–233 °C; [ $\alpha$ ]<sup>25</sup><sub>D</sub> +33.3° (c 3.4, MeOH); FT–IR (dry film)  $\nu_{max}$  3380 (OH), 1700 (C=O) cm<sup>-1</sup>; <sup>1</sup>H NMR ( $C_5D_5N$ , 400 MHz)  $\delta$  0.87 (3H, s, Me-19), 1.00 (3H, s, Me-18), 1.00 (6H, d, J = 6.5 Hz, Me-26 and 27), 1.09 (3H, s, Me-29), 1.33 (3H, s, Me-28), 1.37 (3H, s, Me-30), 2.28 (1H, m, H-25), 2.50 (1H, q, J = 9.9 Hz, H-17), 2.61 (1H, td, J = 11.0, 3.0 Hz, H-20), 4.55 (1H, br s, H-7), 4.88 (1H, d, J = 7.5 Hz, H-1 of glc), 4.88 and 4.91 (each 1H, br s, H<sub>2</sub>-31); <sup>13</sup>C NMR ( $C_5D_5N$ ), see Table 1; FABMS m/z [M – H] – 645, [M – H – glc – H<sub>2</sub>O] – 465; anal. C 68.75%, H 9.18%, calcd for  $C_{37}H_{58}O_9$ , C 68.70%, H 9.04%.

Acid Hydrolysis of Laetiposide A (1). A solution of 1 (20 mg) in 5%  $H_2SO_4$ -dioxane (1:1) was heated at 100 °C for 2 h. The reaction mixture was diluted with H<sub>2</sub>O and extracted with EtOAc. The EtOAc layer was subjected to Si gel column chromatography with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (25:1) to give dehydroeburicoic acid (7, 10 mg) of mp 255–257 °C;  $[\alpha]^{25}_{D}$  +11.5° (c 0.9, pyridine), whose UV, <sup>1</sup>H NMR, <sup>13</sup>C NMR, and EIMS data were consisted with literature values.<sup>5</sup> The aqueous layer was neutralized with Amberlite IRA-35 and evaporated in vacuo to dryness. The identification and the D or L configuration of sugar was determined by using RI detection (Waters 410) and chiral detection (Shodex OR-1) by HPLC [Shodex RSpak NH<sub>2</sub>P-50 4E, CH<sub>3</sub>CN-H<sub>2</sub>O-H<sub>3</sub>PO<sub>4</sub> (95:5:1), 1 mL/min, 47 °C] by comparison with an authentic sugar (10 mmol each of D-glc and L-glc). The sugar portion gave the peak of D-(+)-glc at 20.7 min.

Acid Hydrolysis of Laetiposides B-D (2–4). A solution of each compound 2–4 (each 2 mg) in 5% H<sub>2</sub>SO<sub>4</sub>–dioxane (1: 1) was heated at 100° for 2.0 h. The reaction mixture was diluted with H<sub>2</sub>O and then neutralized with Amberlite IRA-35 and evaporated in vacuo to dryness. The identification of the D or L configuration of the sugar was carried out in the same manner as described for 1 to give the peak of D-(+)-glc at 20.7 min from compounds 2–4.

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